



Pergamon

Tetrahedron Letters 39 (1998) 3333-3336

TETRAHEDRON  
LETTERS

# Diastereoselective Synthesis of 1,2-Diphenyl-1,2-diaminoethanes by Yb(OTf)<sub>3</sub> Accelerated Reductive Coupling of Imines

Rita Annunziata, Maurizio Benaglia, Mauro Cinquini, Franco Cozzi, and  
Laura Raimondi

*Centro CNR and Dipartimento di Chimica Organica e Industriale - Universita' di Milano  
via Golgi 19 - 20133 Milano, Italy*

Received 4 February 1998; accepted 27 February 1998

## Abstract

New reaction protocols have been established to perform the reductive coupling of *N*-benzyl benzaldimines to 1,2-diphenyl-1,2-diaminoethanes in mild, stereoselective, and catalytic conditions by the use of SmI<sub>2</sub> and Yb(OTf)<sub>3</sub>. © 1998 Elsevier Science Ltd. All rights reserved.

**Keywords:** Diamines; Reduction; Stereoselection; Lanthanides.

The importance of the 1,2-diamine functionality in medicinal chemistry [1,2], stereoselective synthesis [3-6], and analytical chemistry [7,8] prompted a great deal of research toward the development of new methods for the synthesis of these compounds [9-11].

Among these methods, the metal-promoted reductive coupling of aldimines, although generally limited to the preparation of symmetrical products, is receiving increasing attention [12-20]. Recently, SmI<sub>2</sub> [21], that has been extensively exploited for the pinacol coupling of carbonyls [22], proved to be a useful promoter of aldimine coupling, provided that either a large excess of reagent [23-25] or high reaction temperatures [23,24] or very activated aldimines [25] were used.<sup>1</sup>

We here report that: i) the combined use of SmI<sub>2</sub> and Yb(OTf)<sub>3</sub> allows the homocoupling of *N*-benzyl benzaldimines to occur at room temperature and in the presence of only 2.0 mol

<sup>1</sup> The system Sm(0)/Cp<sub>2</sub>TiCl<sub>2</sub> has also been used [26] for imine coupling, but it is likely that in this case the reaction is promoted by an *in situ* generated Ti(III) species. For a similar example in carbonyl pinacol coupling see ref. [27].

equiv of  $\text{SmI}_2$  to give the corresponding diamine with low to complete *syn* stereoselectivity; ii) the reaction can be made catalytic in  $\text{SmI}_2$  by the use of a large excess of Mg metal; and iii) when an enantiomerically pure amine is used to generate the imine, a fair level of stereocontrol can be achieved in the coupling reaction.

**Table 1**  
Diastereoselective coupling of aldimines **1a-c** to diamines **2a-c**

Entry	Imine 1	R	Ar	$\text{SmI}_2$ mol equiv	Mg mol equiv	T °C	Diamine 2	Y% <sup>a</sup>	<i>syn/anti</i> <sup>b</sup> ratio
1	<b>a</b>	H	Ph	2.0	0	65	<b>a</b>	53 <sup>c</sup>	57/43
2	<b>a</b>	H	Ph	2.0	0	20	<b>a</b>	81	>98/2
3	<b>a</b>	H	Ph	2.0	0	65	<b>a</b>	84	63/37
4	<b>a</b>	H	Ph	2.0	0	-20	<b>a</b>	83	80/20
5	<b>a</b>	H	Ph	2.0	8	65	<b>a</b>	96 <sup>c</sup>	50/50
6	<b>a</b>	H	Ph	0.2	8	65	<b>a</b>	42 <sup>c</sup>	56/44
7	<b>a</b>	H	Ph	0.2	8	65	<b>a</b>	62	63/37
8	<b>b</b>	Me	Ph	2.0	0	20	<b>b</b>	86	62/38 <sup>d</sup>
9	<b>b</b>	Me	Ph	2.0	0	-20	<b>b</b>	85	64/36 <sup>e</sup>
10	<b>c</b>	Me	$\alpha$ -Napht	2.0	0	20	<b>c</b>	23	65/35 <sup>f</sup>

<sup>a</sup> Isolated yield after flash chromatography. <sup>b</sup> As determined by 300 MHz  $^1\text{H}$  NMR analysis of the crude products.

<sup>c</sup> In the absence of  $\text{Yb}(\text{OTf})_3$ . <sup>d</sup> Two *syn* isomers were obtained in a 75/25 ratio. <sup>e</sup> Two *syn* isomers were obtained in a 89/11 ratio. <sup>f</sup> A single *syn* isomer was detected. For **2b** and **2c**, only the major *syn* isomer is depicted.

As can be seen from the data collected in Table 1 the coupling reaction of imine **1a** promoted by  $\text{SmI}_2$  occurred in *refluxing* THF to afford a 53/47 mixture of *syn* and *anti* isomers of diamine **2a** (entry 1).<sup>2</sup> No reaction was observed at room temperature. On the basis of the recognized [29-32] ability of  $\text{Yb}(\text{OTf})_3$  to activate imines toward nucleophilic attack at carbon, we attempted the coupling reaction in the presence of 1 mol equiv of this Lewis acid.<sup>3</sup> We found that not only the reaction occurred at room temperature, but also the *syn* diastereoselectivity was increased to >98/2 (entry 2). Running the reaction at -20 or at 65°C (entries 3 and 4) depressed the diastereoselectivity at different extent.

Prompted by a recent report describing a  $\text{SmI}_2$  catalyzed pinacol coupling of aldehydes [33]

<sup>2</sup> Configurational assignment resided on comparison of  $^1\text{H}$  NMR data of the isomers of compound **2a** with those reported [28].

<sup>3</sup> Also the use of  $\text{BF}_3\text{OEt}_2$  allowed to run the reaction at room temperature, but the diastereoselection was lower, **2a** being obtained in 67% yield as a 63/37 *syn/anti* mixture of diastereoisomers.

we also realized a catalytic version of our imine coupling process. To this end, the reaction of imine **1a** was carried out in refluxing THF in the presence of 0.2 mol equiv of  $\text{SmI}_2$  and of 8 mol equiv of Mg metal to afford a 56/44 mixture of *syn* and *anti* isomers of diamine **2a** in 42% yield (entries 5 and 6). Addition of  $\text{Yb}(\text{OTf})_3$  (entry 7) did not allow to lower the reaction temperature but increased both the yield (up to 62%) and, slightly, the diastereoselection (*syn/anti* = 63/37).

Finally, control of the absolute stereochemistry of the products was attempted by using an enantiomerically pure residue at the imine nitrogen. When imine (R)-**1b** was coupled in the presence of  $\text{SmI}_2$  (2.0 mol equiv) and  $\text{Yb}(\text{OTf})_3$  (1.0 mol equiv) at room temperature (entry 8), a 68/32 mixture of two *syn* and one *anti* isomers was obtained in 86% yield. The major *syn* isomer had the (R,R,R,R) configuration and was obtained in a 75/25 excess over the (R,S,S,R)-one [34].<sup>4</sup> The ratio between the *syn* isomers was increased to 89/11 by lowering the temperature down to  $-20^\circ\text{C}$  (entry 9). Replacement of the (R)-1-phenylethanamine residue of **1b** by the bulkier (R)-1- $\alpha$ -naphthylethanamine one of **1c** led to a higher stereocontrol even at room temperature (a single *syn* isomer of **2c** was obtained), but also to a marked decrease of the chemical yield (entry 10).

In conclusion, new reaction protocols have been established that allow efficient imine reductive homocoupling to generate 1,2-diphenyl-1,2-diaminoethanes in mild, stereoselective, and catalytic conditions.<sup>5</sup> Work is in progress to extend this reaction to the stereocontrolled preparation of other vicinal diamines.<sup>6</sup>

**Acknowledgement.** Partial financial support from MURST and CNR is gratefully acknowledged.

#### References

- [1]. Michalson ET, Szmuszkovicz J. *Progr. Drug. Res.* 1989;33:135-149.
- [2]. Reedijk J. J. *Chem. Soc., Chem. Commun.* 1996:801-806.
- [3]. Katsuki T. *Coord. Chem. Rev.* 1995;140:189-214.
- [4]. Jacobsen EN. *Asymmetric Catalytic Epoxidation of Unfunctionalized Olefins*. In: Ojima I, editor. *Catalytic Asymmetric Synthesis*. New York: VCH, 1993: 159-202.
- [5]. Togni A, Venanzi LM. *Angew. Chem. Int. Ed. Engl.* 1994;33:497-526.

<sup>4</sup> *Syn* and *anti* isomers are easily distinguished by  $^1\text{H}$  NMR spectroscopy on the basis of the Ph-CH-NH-CH(Me)Ph signals: these are homotopic and isochronous in the  $C_2$  symmetric *syn* isomers, and diastereotopic and anisochronous in the  $C_1$  symmetric *anti* isomer. The  $^1\text{H}$  NMR spectrum of the (R,S,S,R) isomer of **2b** reported in ref. [33] is identical to that of the minor component of our mixture. Therefore, the major *syn* isomer has the (R,R,R,R) configuration. For a very recent application of the lithium salt of diamine *syn* **2b** to stereoselective synthesis see ref. [35].

<sup>5</sup> Commercially available 0.1 M solutions of  $\text{SmI}_2$  in THF were employed throughout this work. To ensure reproducible results, a freshly opened bottle of reagent must be used. *In situ* generation of  $\text{SmI}_2$  from  $\text{CH}_2\text{I}_2$  and Sm metal led to similar but less reproducible results. The imines were freshly prepared and aldehyde free (by NMR) compounds that were used as crude products.

<sup>6</sup> In ancillary experiments it was found that also the following imines could be coupled in the conditions of entry 2, Table 1: *N*-benzyl 4-fluorobenzaldimine (95% yield, *syn/anti* = 60/40); *N*-benzyl 4-methoxybenzaldimine (28% yield, *syn/anti* = 70/30); *N*-*t*-butyl benzaldimine (80% yield, *syn/anti* = 65/35).

- [6]. Cox PJ, Simpkins NS. *Tetrahedron:Asymmetry*. 1991;2:1-26.
- [7]. Alexakis A, Frutos JC, Mutti S, Mengeny P. *J. Org. Chem.* 1994;59:3326-3334;and references therein.
- [8]. Filwood R, Parker D. *J. Chem. Soc. Perkin Trans. 2*. 1994:57-64.
- [9]. Enders D, Wiedemann J. *Synthesis*;1996: 1443-1450; and references therein.
- [10]. Enders D, Chelain E, Raabe G. *Bull Soc. Chim. Fr.* 1997;134:299-306.
- [11]. Merino P, Lanaspá A, Merchan FL, Tejero T. *Tetrahedron:Asymmetry*. 1997;8:2381-2401; and references 8-11 therein.
- [12]. Alkali-earth metals: Smith JG, Ho I. *J. Org. Chem.* 1972;37:653-656;and references therein.
- [13]. Aluminum: Beruah B, Prayapati D, Sandhu JS. *Tetrahedron Lett.* 1995;36:6747-6750; and references therein.
- [14]. Zinc: Shimizu M, Iida T, Fujisawa T. *Chemistry Lett.* 1995:609-610; and references therein.
- [15]. Zirconium: Buchwald SL, Watson BT, Wannamaker MW, Dewan JC. *J. Am. Chem. Soc.* 1989;111:4486-4494.
- [16]. Indium: Kalyanam N, Venkateswara Rao G. *Tetrahedron Lett.* 1993;34:1647-1648.
- [17]. Ytterbium: Takuki K, Tsukabi Y, Tanaka S, Beppu F, Fujiwara Y. *Chemistry Lett.* 1990:203-204.
- [18]. Vanadium: Imwinkelried R, Seebach D. *Helv. Chim. Acta.* 1984;67:1496-1502.
- [19]. Niobium: Roskamp EJ, Pedersen SF. *J. Am. Chem. Soc.* 1987;109:3152-3154.
- [20]. Titanium: Betschart C, Schmidt B, Seebach D. *Helv. Chim. Acta.* 1988;71:1999-2021; and references therein.
- [21]. Molander GA, Harris CR. *Chem. Rev.* 1996;96:307-338.
- [22]. Robertson GM. Pinacol Coupling Reaction. In: Trost, BM, Fleming, I, editors. *Comprehensive Organic Synthesis*. Oxford: Pergamon Press, 1991;vol.3:563-611.
- [23]. Enholm EJ, Forbes DC, Holub DP. *Synthetic Commun.* 1990;20:981-987.
- [24]. Imamoto T, Nishimura S. *Chemistry Lett.* 1990:1141-1142.
- [25]. Taniguchi N, Uemura M. *SYNLETT*. 1997:51-53
- [26]. Liao P, Huang Y, Zhang Y. *Synthetic Commun.* 1997;27:1483-1486.
- [27]. Gansäuer A. *SYNLETT*. 1997:363-364.
- [28]. Tanaka H, Dhimane H, Fujita H, Ikemoto Y, Torii S. *Tetrahedron Lett.* 1988;29:3811-3814.
- [29]. Kobayashi S, Ishitani H, Nagayama S. *Synthesis* 1995:1195-1202.
- [30]. Kobayashi S, Nagayama S. *J. Am. Chem. Soc.* 1997;119:10049-10053; and references therein.
- [31]. Annunziata R, Cinquini M, Cozzi F, Molteni V, Schupp O. *J. Org. Chem.* 1996;61:8293-8296.
- [32]. Annunziata R, Benaglia M, Cinquini M, Cozzi F, Molteni V, Schupp O. *Tetrahedron* 1997;53:9715-9726.
- [33]. Nomura R, Matsuno T, Endo T. *J. Am. Chem. Soc.* 1996;118:11166-11167.
- [34]. Bambridge K, Begley MJ, Simpkins NS. *Tetrahedron Lett.* 1994;35:3391-3394.
- [35]. Gibson SE, Ham P, Jefferson GR. *J. Chem. Soc., Chem. Commun.* 1998:123-124.